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54 Improved mineral filled rubber composites.

57 A reinforced rubber composite which comprises a vulcanized, or vulcanizeable, rubber matrix material, mineral filler dispersed in the rubber matrix material, and a mixture of at least two silanes where the first silane - (a) contains least one alkenyl radical, and the second silane (b) contains at least one radical with aromatic character such as a phenyl radical. The invention also relates to mineral fillers treated with said mixture of at least two silanes.

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IMPROVED MINERAL FILLED RUBBER COMPOSITES

This invention relates to mineral reinforced rubber used in electrical applications as insulating materials, wherein the mineral filler has been treated prior to being compounded with the rubber with a blend of at least two silanes.

Rubber is widely used as an electrical insulating material, and filled rubber has also been used for those purposes. Filled rubber is a composite material comprised of cured rubber and filler incorporated throughout the cured rubber.

In general, two types of fillers are used to reinforce or fill rubber. Carbon black fillers are widely used as reinforcing agents and greatly improve the strength of the rubber composites in which they are incorporated. Nonblack fillers are the fillers other than carbon black. Typically, the nonblack fillers are mineral fillers which require some form of treatment to act as reinforcement in the composites in which they are incorporated. Mineral filled or reinforced rubber is often used instead of rubber because of reduced cost and improved electrical and physical properties over unfilled rubber.

Silanes have been used for a number of years to improve the physical performance of mineral filled or reinforced rubber materials. Some of the improvements imparted by the use of silanes to the filled rubber include improved hardness, flexibility, elasticity, and resistance to weathering. The use of silanes in mineral filled, or reinforced rubber systems is analogous to the use of silanes as coupling agents in other composites, such as laminates. However, because of the elasticity of mineral filled rubber, such systems do not behave precisely as other composites in regards to the optimal choice of silane coupling agents is concerned.

For instance, U.S. Patent No. 2,742,378 issued to Grotenhuis teaches filler materials treated with "vinylsiloxane" groups impart increased flexural strength to polymers filled with said treated fillers. U.S. Patent No. 2,763,629 issued to Gottfurcht discloses glass fibers treated with triethoxyvinylsilane impart improved flexural strength and resistance to hydrolytic degradation of flexural strength when incorporated into polyester resin laminates. U.S. Patent No. 3,013,915 issued to Morgan teaches vinylsilanes effectively treat glass fibers so that polyolefins filled with the treated glass fibers have improved physical properties and weatherability.

The use of silanes in filled electrical insulating materials is disclosed in U.S. Patent No. 3,533,348 issued to Betts. Betts specifically discloses an electrical insulating composition for coating wire and cable comprising cross-linked ethylene copolymer, chlorine containing polymer (for fire resistance), and magnesium silicate filler treated with alkoxysilane. One of the alkoxy silanes particularly taught is vinyltris-(2-methoxyethoxy)silane.

Blends of different silanes are used as coupling agents for glass reinforced laminates. For instance, U.S. Patent No. 3,317,369 issued to Plueddemann teaches that a blend of silanes comprised of acryloxyalkyl substituted silanes of the formula $RSiX_n$, where R denotes an acryloxyalkyl radical and X represents a hydrolyzable radical, and methyltrimethoxyphenylsilane gives improved flexural strength to glass fiber reinforced polyester laminates.

Ashcraft in U.S. Patent No. 4,144,202 teaches that water treeing in ethylene polymers can be inhibited by the addition of 0.5 to 2 percent epoxy functional silanes to the unfilled polymers. U.S. Patent No. 4,229,713 issued to Maringer discloses the use of silane blends as anti-treeing additives in unfilled polymeric compositions. Specifically, Maringer teaches incorporating 2 weight percent of vinyl-tris-(methoxyethoxy)silane in polyethylene inhibits electrical treeing.

The mineral filler in reinforced rubber is often treated with a silane coupling agent to improve the electrical and physical properties of the material. Generally, silane treated fillers produce high quality reinforced rubber. However, it has been found that the electrical insulating properties of such reinforced rubber degrade upon exposure to water, or moisture. A mineral reinforced rubber which showed increased weatherability over standard reinforced rubber would be useful in electrical insulation applications which subject the electrical components to adverse weather and moisture conditions.

This invention is a mineral filler reinforced rubber composite which comprises a rubber matrix material, a mineral filler dispersed in the rubber matrix which has been treated with a blend of at least two silanes where the first silane (a) is of the general formula



where R denotes a vinyl or allyl radical which can react chemically with the rubber matrix material, X denotes a hydrolyzable radical such as alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy, and n is 1 or 2; and

the second silane (b) is of the general formula



where R' denotes an aromatic radical such as phenyl, benzyl, naphthyl, or the like, Y denotes a hydrolyzable radical such as alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy, and m equals 1, or 2.

5 The mineral filler reinforced rubber composites of the invention can be used to coat wires, and to encapsulate electronic components.

The invention is a reinforced rubber which comprises a vulcanized, or vulcanizeable, rubber matrix material, mineral filler dispersed in the rubber matrix material, and a mixture of at least two silanes where the first silane (a) contains at least one alkenyl radical, and the second silane (b) contains at least one
10 radical with aromatic character such as a phenyl radical. The invention also relates to mineral fillers treated with said mixture of at least two silanes.

The vulcanized or vulcanizeable rubber matrix material encompassed within the invention includes ethylene-propylene rubber (EPR), ethylene-propylene terpolymer rubber (EPDM), cross-linked polyethylene rubber, and silicone rubber. These types of rubber are commercially available, and can be vulcanized by a
15 variety of methods well known in the prior art. The silanes of the invention interact with the rubber matrix material most effectively during the vulcanization of the rubber matrix. Preferably, the rubber matrix material is cured subsequent to the incorporation of the mineral filler and silane blend with the rubber matrix material.

The mineral fillers encompassed by the invention include, but are not limited to, calcined clay, titanium
20 dioxide, hydrated silica, hydrated sodium silicate, magnesium carbonate, kaolin, aluminum hydroxide - (bauxite), basic aluminum sulfate, precipitated calcium carbonate, alumina silicate, hydrous alumina silicate, and other nonblack fillers. Preferably, commercially available fillers are used which are compatible with the rubber matrix material, and which have the particle size and structure best suited for acting as reinforcement of the rubber matrix material.

25 The silane blends used in the invention are mixtures of at least two silanes; an alkenyl radical containing silane (a) and an aromatic radical containing silane (b).

The alkenyl containing silanes (a) encompassed by the invention are of the general formula



where R denotes a vinyl or allyl radical which can chemically react with the matrix material and remain
30 chemically bonded to the silicon atom, X denotes a hydrolyzable radical, and n is 1 or 2. The R radical chosen will depend upon the character of the rubber matrix material being employed. For instance, in the case of an ethylene-propylene rubber, EPR, vinyl is preferred due to its high reactivity and commercial availability. X includes, but is not necessarily limited to, alkoxy radicals containing 1 to 5 carbon atoms, acyloxy radicals with 2 to 5 carbon atoms, halo, or alkoxyalkoxy radicals which contain 2 to 10 carbon
35 atoms.

Specific alkenylsilanes within the limits of the invention are trimethoxyvinylsilane, triacetoxymethylsilane, tris-(2-methoxyethoxy)vinylsilane, allyltrimethoxysilane, trichlorovinylsilane, tripropoxyvinylsilane, tributoxymethylsilane, tributyltrimethoxysilane, alkyltriisovaleryloxysilane, or tris(2-ethoxypropoxy)vinylsilane.

The aromatic-radical containing silane (b) is of the general formula



where R' denotes an aromatic radical. For example, R' could be a phenyl, tolyl, naphthyl, or like radicals. Preferably, R' is chosen so that the silane is commercially available and usually will be phenyl. Y denotes a hydrolyzable radical as defined for the alkenylsilane, ie. Y denotes alkoxy, acyloxy, alkoxyalkoxy, halo, or the like. m is 1 or 2. Preferably, m is 1. Specific silanes include phenyltrimethoxysilane, dimethoxydiphenyl-
45 silane, trichlorophenylsilane, tributoxyphenylsilane, triisopentoxybenzylsilane, phenyltripropionyloxysilane, naphthyltrivaleryloxysilane, and tris(2-methoxyethoxy)phenylsilane.

The ratio of (b) to (a) will depend upon the nature of the rubber matrix, the effectiveness of the alkenyl containing silane as a coupling agent, and the amount of silane blend being used to treat the mineral filler. Preferably, the aromatic radical containing silane should comprise between 25 to 85 weight percent of the
50 silane blend. Most preferably, the aromatic radical containing silane provides at least 60 percent of the silane blend by weight.

The silane blends should be employed in levels from 0.2 to 3 weight percent of the mineral filler material.

The proportion of the components of the invention can vary. The mineral filler should comprise between
55 30 and 200 parts by weight per 100 parts by weight of the rubber matrix material. The amount of silane blend employed varies according to the weight of mineral filler used in the particular embodiment, but typically comprises less than 1 weight percent of the reinforced rubber composite.

The mineral reinforced rubber composites of the invention can be made by a variety of methods. The filler can be treated with the silane blend before being incorporated into the rubber matrix material, or the silane blend can be added to a mixture of the uncured rubber matrix material and mineral filler so the filler is treated in situ. The mineral filler can be treated with the silane blends several ways. For instance, the silane blend can be mixed directly with the filler, or, alternately, the silane blend can be mixed with organic solvent or water to form a slurry which can be added mixed with filler.

The improved electrical properties of the invention are believed to be attributable to the ability of the aromatic radical containing silane to bear electrical loads. Both the alkenylsilane and aromatic radical containing silane interact with the surface of the mineral filler, so a relatively high concentration of both silanes exists in the interfacial region between the filler and matrix of the reinforced rubber. The electrical load bearing capacity of the aromatic radical containing silanes of the invention in conjunction with the coupling agent activity of the alkenylsilane improves the electrical properties of the reinforced rubber at the interfacial portion between the mineral filler and rubber matrix material of the reinforced rubber. The exact reason for the improved electrical properties is not understood, and this explanation is not intended to limit the scope of the invention which is defined in the claims.

The following example illustrates the invention, but does not fully delineate the full scope of the invention. An example outside the scope of the invention is included for comparison.

20 EXAMPLE 1

Calcined clay was treated with a blend of silanes comprised of 60 wt. % phenyltrimethoxysilane and 40 wt. % triacetoxysilane. The silane blend was added neat to the calcined clay at 1 wt. %, the mixture was mixed to disperse the silane throughout the clay, and the mixture was dried. The treated clay was compounded with an ethylene-propylene rubber to form a mixture comprised of 100 parts of an ethylene-propylene rubber gum (Vistalon 404 EPR sold by Huber Corp., Macon, Georgia), 75 parts of the treated clay, 5 parts of zinc oxide, 0.5 part of the antioxidant Agerite Resin D (R.T. Vanderbilt Co., Inc., N.Y., N.Y.), and 10 parts of dicumyl peroxide (Di Cup 40C, The Hercules Powder Co., Inc., Wilmington, Delaware). The mixture was compounded into two slabs (slab A and slab A'). Slab A was vulcanized at 330°C for 25 minutes, and slab A' was cured at 330°C for 50 minutes.

Slabs B and B' were made in the same manner except the silane blend was replaced with an equivalent weight of triacetoxysilane. Slab B was cured at 330°C for 25 minutes, slab B' for 50 minutes. Slabs C and C' were made using the same methods except the silane blend comprised 70 wt. % phenyltrimethoxysilane, and 30 wt. % triacetoxysilane.

All slabs were characterized with respect to their physical and electrical properties using standard ASTM test methods. Particularly, the hardness, 100% moduli, 200% moduli, 300% moduli, tensile strength, and elongation of each slab were measured. These results are summarized in the following table.

PHYSICAL PROPERTIES OF CLAY REINFORCED RUBBER SLABS

	<u>A/A'</u>	<u>B/B'</u>	<u>C/C'</u>
100% Modulus, psi	280/280	230/240	290/270
200% Modulus, psi	590/560	570/570	570/500
300% Modulus, psi	740/760	810/860	710/760
tensile strength, psi	810/810	870/850	800/770
elongation, %	310/350	340/330	380/400
hardness (Shore A)	58/58	58/59	58/58

The physical properties of slabs A, A', C, and C' are similar to the physical properties of slabs B and B' which reflect the state of the art in treating clay used in reinforced rubber.

The electrical properties of slabs A', B', and C', which were cured for 50 minutes, are reported in the following table. The dielectric constant and dissipation factor were determined using the methods described in ASTM D 150. The volume resistivity was determined using the method of ASTM D 257.

ELECTRICAL PROPERTIES OF
TREATED CLAY REINFORCED RUBBER SLABS

	<u>A'</u>	<u>B'</u>	<u>C'</u>
5 Dielectric Constant, 10*2 Hz	2.6	2.6	2.6
10 Dielectric Constant, 10*5 Hz	2.5	2.5	2.5
Dissipation Fact. 10*2 Hz	4.2×10^{-3}	3.7×10^{-3}	4.0×10^{-3}
15 Dissipation Fact. 10*5 Hz	2.6×10^{-3}	3.0×10^{-3}	3.6×10^{-3}
Volume Resistivity (ohm/cm.)	3.1×10^{15}	1.7×10^{15}	1.4×10^{15}

20 The values in the table represent the measured value of each slab under dry conditions. The slabs made with the calcined clay treated with the silane blend show roughly similar dry electrical properties. The three slabs were immersed in water for 1 day and their electrical properties were remeasured.

ELECTRICAL PROPERTIES AFTER 1 DAY H₂O IMMERSION

	<u>A'</u>	<u>B'</u>	<u>C'</u>
25 Dielectric Constant, 10*2 Hz	2.7	2.6	2.6
30 Dielectric Constant, 10*5 Hz	2.6	2.5	2.5
Dissipation Fact. 10*2 Hz	20×10^{-3}	13×10^{-3}	7.7×10^{-3}
35 Dissipation Fact. 10*5 Hz	4.9×10^{-3}	5.6×10^{-3}	4.8×10^{-3}
Volume Resistivity	1.0×10^{15}	0.46×10^{15}	3.2×10^{15}

40 The slabs using the silane blends of the invention displayed less degradation of volume resistivity than the comparative example B'. The three slabs were immersed for six more days and the electrical properties were remeasured.

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ELECTRICAL PROPERTIES AFTER 7 DAY H₂O IMMERSION

	<u>A'</u>	<u>B'</u>	<u>C'</u>
5 Dielectric Constant, 10*2 Hz	2.7	3.1	2.7
Dielectric Constant, 10*5 Hz	2.6	2.5	2.6
10 Dissipation Fact. 10*2 Hz	25×10^{-3}	132×10^{-3}	59×10^{-3}
Dissipation Fact. 10*5 Hz	7.2×10^{-3}	5.6×10^{-3}	17×10^{-3}
15 Volume Resistivity	1.1×10^{15}	0.13×10^{15}	0.65×10^{15}

The standard rubber composite, B', showed marked degradation of its volume resistivity from the dry measurements. An overall loss of greater than 90 percent was observed in slab B', whereas the slabs encompassed in the invention, A' and C', showed volume resistivity losses of less than 70 percent from their dry volume resistivities. Similar resistance to the degradative affects of water immersion are displayed in the dissipation factors of slabs A' and C' at 10²Hz.

25 Claims

1. A reinforced rubber composite which comprises;
 - (I) a rubber matrix material;
 - (II) a mineral filler incorporated in the rubber matrix material; and
 - (III) a blend of at least two silanes, where the first silane (a) is of the general formula R_nSiX_{4-n} where R denotes a vinyl or allyl radical, X denotes an alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy radical, and n is 1 or 2; and the second silane (b) is of the general formula R'_mSiY_{4-m} where R' denotes a phenyl, benzyl, tolyl, or naphthyl radical, Y denotes an alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy radical, and m equals 1, or 2.
2. The reinforced rubber composite of claim 1 wherein said composite comprises between 30 and 200 parts by weight of the mineral filler (II) and between 0.06 and 6 parts by weight of the silane blend per 100 parts by weight of the rubber matrix material (I).
3. The reinforced rubber composite of claim 1 wherein the rubber matrix material is chosen from the group consisting of ethylene propylene rubber, crosslinked polyethylene rubber, ethylene propylene terpolymer rubber, or silicone rubber.
4. The reinforced rubber composite of claim 1 wherein (b) comprises between 25 and 85 weight percent of the silane blend (III).
5. The reinforced rubber composite of claim 4 wherein (b) comprises at least 60 weight percent of the silane blend (III).
6. The reinforced rubber composite of claim 5 wherein the first silane (a) is triacetoxymethylsilane, and the second silane (b) is phenyltrimethoxysilane.
7. The reinforced rubber composite of claim 6 wherein the rubber matrix material is an ethylene-propylene rubber, and the mineral filler is calcined clay.
8. Treated mineral filler which comprises;
 - (I) mineral filler; and
 - (II) a blend of at least two silanes, where the first silane (a) is of the general formula R_nSiX_{4-n} where R denotes a vinyl or allyl radical, X denotes an alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy radical, and n is 1 or 2; and the second silane (b) is of the general formula

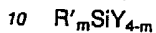


where R' denotes a phenyl, benzyl, or naphthyl radical, Y denotes an alkoxy, acyloxy, halo, hydroxy or alkoxyalkoxy radical, and m equals 1 or 2.

9. A method for making mineral filler reinforced rubber with improved electrical properties which comprises; mixing an uncured rubber matrix material, a mineral filler, and a mixture comprised of at least two silanes, where the first silane (a) is of the general formula



where R denotes a vinyl or allyl radical, X denotes an alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy radical, and n is 1 or 2; and the second silane (b) is of the general formula



where R' denotes a phenyl, benzyl, or naphthyl radical, Y denotes an alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy radical and m equals 1, or 2, and curing the mixture.

10. A mineral filler treating composition consisting essentially of;
a silane (a) of the general formula



where R denotes a vinyl or allyl radical, X denotes an alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy radical, and n is 1 or 2; and

a silane (a) of the general formula



- where R' denotes a phenyl, benzyl, or naphthyl radical, Y denotes an alkoxy, acyloxy, halo, hydroxy, or alkoxyalkoxy radical and m equals 1, or 2, and (b) comprises between 60 and 85 weight percent of the composition.

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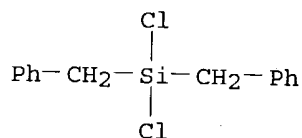
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C.A. 43, 2036a.

IT 18414-36-3, Silane, dibenzylchloro-
(siloxanes from)
RN 18414-36-3 CAPLUS
CN Silane, dichlorobis(phenylmethyl)- (9CI) (CA INDEX NAME)



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TI **Organosiloxanes**

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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AB **Siloxane** polymers are prepared by cohydrolyzing a mixture of at least 2 silanes containing 2 univalent hydrocarbon radicals (aryl, alkyl, or alkenyl) which need not be the same, attached to each Si atom through C-Si linkages, the remaining valences of the Si being hydrolyzable radicals (attached by other than C-Si linkages (Cl, OR). To a mixture of 224 g. of (C₆H₅CH₂)MeSi(OEt)₂ and 296 g. of Me₂Si(OEt)₂, 1325 cc. of 95% EtOH containing a few drops of concentrated HCl was added slowly with warming to effect hydrolysis and intercondensation. H₂O 25 cc. was added and the solvent removed by evaporation. A very viscous oil was obtained which formed a waxy substance upon heating at 190° for 48 hrs. Other compds. used are Ph₂SiCl₂, PhEtSiCl₂, (PhCH₂)₂SiCl₂, and PhMeSi(OEt)₂.

IT 18414-36-3, Silane, dibenzylchloro-
(siloxanes from)

RN 18414-36-3 CAPLUS

CN Silane, dichlorobis(phenylmethyl)- (9CI) (CA INDEX NAME)

